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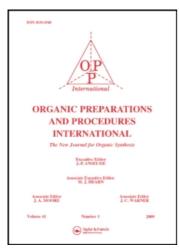
On: 27 January 2011

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# Organic Preparations and Procedures International

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t902189982

## THE PREPARATION OF γ-THIOPYRAN

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To cite this Article Strating, J. and Molenaar, E.(1969) 'THE PREPARATION OF  $\gamma$ -THIOPYRAN', Organic Preparations and Procedures International, 1: 1, 21 - 23

To link to this Article: DOI: 10.1080/00304946909458341 URL: http://dx.doi.org/10.1080/00304946909458341

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#### THE PREPARATION OF Y-THIOPYRAN

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In a communication by Strating, Keijer, Molenaar and Brandsma<sup>1</sup> the preparation of  $\gamma$ -pyran and  $\gamma$ -thiopyran was announced briefly. Whereas the detailed procedure for the synthesis of 4 H-pyran ( $\gamma$ -pyran) has been given<sup>2</sup>, the experimental details of the synthesis of 4 H-thiopyran ( $\gamma$ -thiopyran) are only reported in the thesis of E. Molenaar<sup>3</sup>.

#### EXPERIMENTAL

To 500 ml of dichloromethane, cooled to  $-70^{\circ}$ , 100 g (1.0 mole) of glutaric dialdehyde, obtained by heating the polymer to about  $110^{\circ}$  \*\*, was added. A mixture of gaseous HCl and H<sub>2</sub>S in the ratio of about 2:1 \*\*\* was passed in the above stirred solution. The temperature of the reaction mixture was kept between -30 and -25° by means of a dry ice acetone bath. After about 3 hours, the introduction of gas was stopped and the

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temperature of the reaction mixture raised to -15° by removing the bath after which, without stirring, the mixture was cooled slowly to  $-70^{\circ}$ . The crystals of ice were removed by filtration and washed with cold dichloromethane  $(-70^{\circ})$ . The filtrate was stirred with powdered calcium chloride until room temperature was reached and then filtered through a fluted filter paper. The solution was evaporated under reduced pressure (the receiver was cooled to  $-80^{\circ}$ ). The residue (which may contain some solvent) was gradually heated to  $135^{\circ}$  together with 300 g (2.0 mole) of N,N-diethylaniline. The entire operation was carried out with stirring in a nitrogen atmosphere \*\*. After the mixture had been cooled to 100°, the crude reaction product was removed by distillation under reduced pressure (receiver cooled to  $-80^{\circ}$ ). The liquid boiling from  $40-80^{\circ}/16$  mm was collected. Redistillation through a Widmer column gave γ-thiopyran with a purity of 98% (determined by GLC, Reoplex on Chromosorb W.A.W., temp. 100°, He flow 50 ml/min.). The boiling point of this product was  $30^{\circ}/12$  mm, the melting point  $-28^{\circ}$ ,  $n_{D}^{20}$  1.5623. The yield varied from 20-45%.  $\gamma$ -Thiopyran is a colorless, clear liquid with a penetrating odor; at room temperature and in contact with air it decomposes rapidly. In the solid state it can be stored for a long time.

IR spectrum (liquid film; Leitz spectrophotometer): 3100 cm<sup>-1</sup> (= CH), 2940 and 2880 cm<sup>-1</sup> (- CH<sub>2</sub> $\dot{-}$ ), 1640 cm<sup>-1</sup> (C=C), 1600 cm<sup>-1</sup> (C=C-S?), 1450 and 1360 cm<sup>-1</sup> (CH<sub>2</sub> def.) and 750 cm<sup>-1</sup> (C-S?). Unidentified peaks: 1320 cm<sup>-1</sup> (s), 1040 cm<sup>-1</sup> (m), 1000 cm<sup>-1</sup> (m), 980 cm<sup>-1</sup> (s), 930 cm<sup>-1</sup> (m) and 850 cm<sup>-1</sup> (broad, s).

UV spectrum (petroleum ether solution; Beckmann DK 2 spectrophotometer): maximum at 278 m $\mu$  ( $\epsilon$  = 2430, log  $\epsilon$  = 3.34) and a shoulder at 236 - 238 m $\mu$  ( $\epsilon$  = 5270, log  $\epsilon$  = 3.72).

PMR spectrum (in CCl<sub>4</sub> at -40°): S-CH=C ( $\tau$  4.1, very complex), S-C=CH ( $\tau$  4.5, very complex) and - CH<sub>2</sub>- (triplet with J = 4 cps; each peak consists of a triplet with J = 1 cps,  $\tau$  = 7.15); integration ratio 1:1:1.

Analysis: Found: C 61.5, 61.4; H 6.3, 6.3; S 32.8, 32.7.

C<sub>5</sub>H<sub>6</sub>S (98.16) Calc.: C 61.17 ; H 6.16 ; S 32.67.

#### REFERENCES

- 1) J. Strating, J. H. Keijer, E. Molenaar and L. Brandsma, Angew. Chem., Intern. Ed. I, 399 (1962).
- Houben Weyl, "Methoden der Organischen Chemie", VI/4, p. 108, 1966.
- 3) E. Molenaar, Thesis Groningen, 1967.
- \* By distilling a commercially available 25-30% solution of glutaric dialdehyde in water, through a 30 cm Widmer column, water is removed at 25°/15 mm. After the removal of water, glutaric dialdehyde distills at 76°/12 mm. At room temperature even when kept under nitrogen, it polymerises rapidly to a syrupy mass.

  Depolymerisation of the glutaric dialdehyde polymer is effected by heating it. When the temperature reaches approximately 85°, depolymerisation appears to be complete and the temperature which previously increased slowly, now rises rapidly. Heating is discontinued at about 110°. The liquid may then be used without distillation.
- \*\* The underlined parts are particularly important for obtaining the yields mentioned.
- \*\*\* The approximate ratio of 2:1 was obtained by comparing the velocity of bubbling through two washing flasks each filled with the same quantity of paraffin oil.

(Received June 21, 1968)